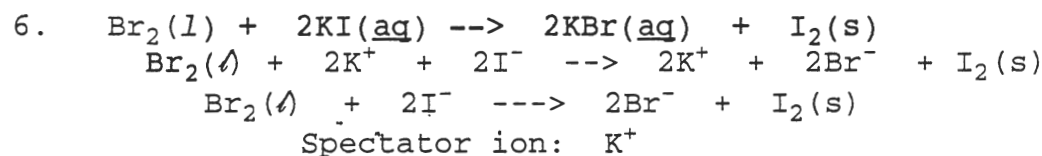
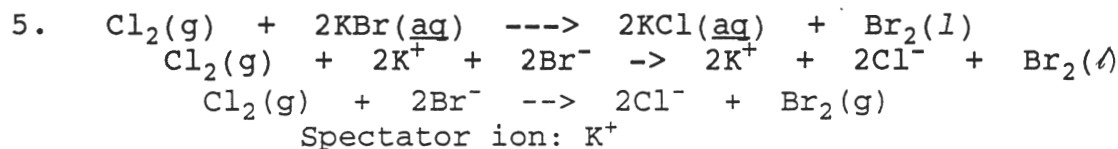
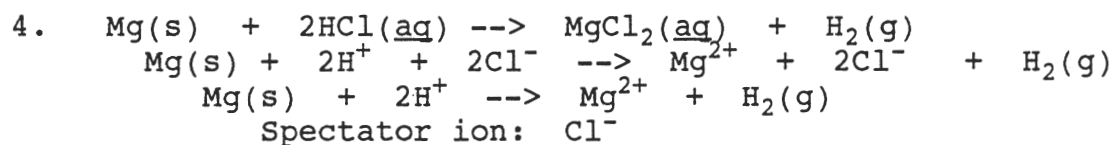
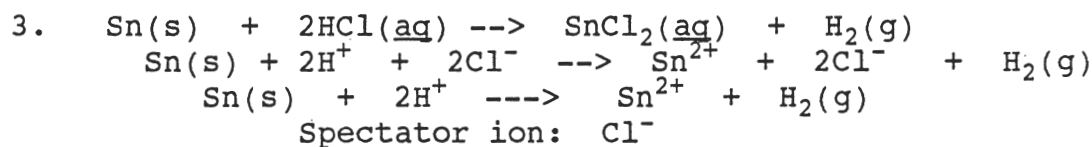
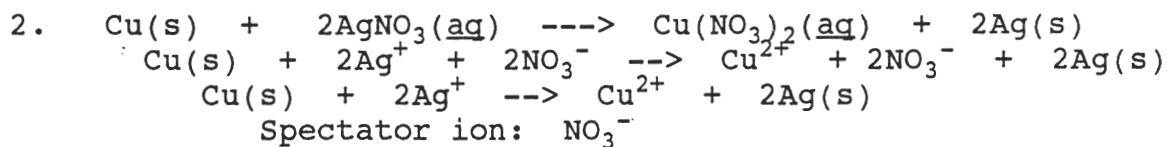
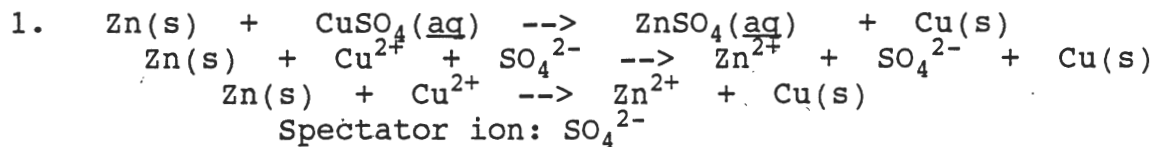


Name: \_\_\_\_\_

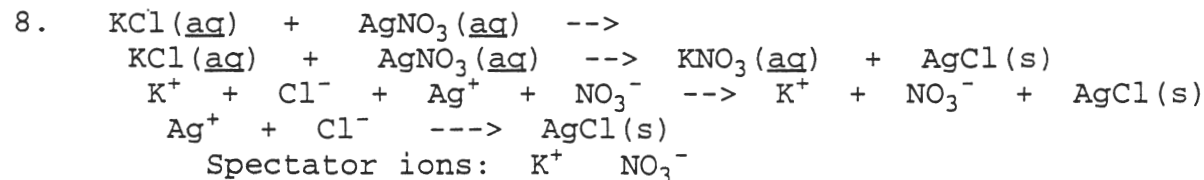
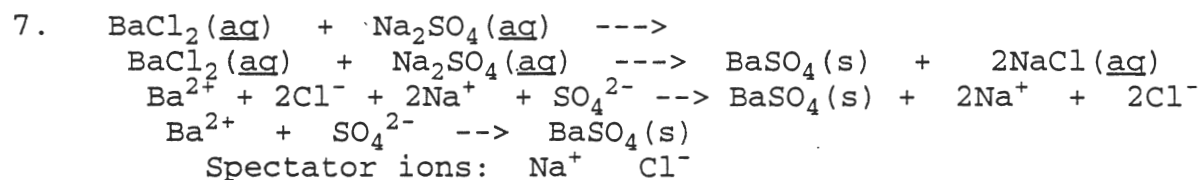
Net Ionic Equations: For each of the following:

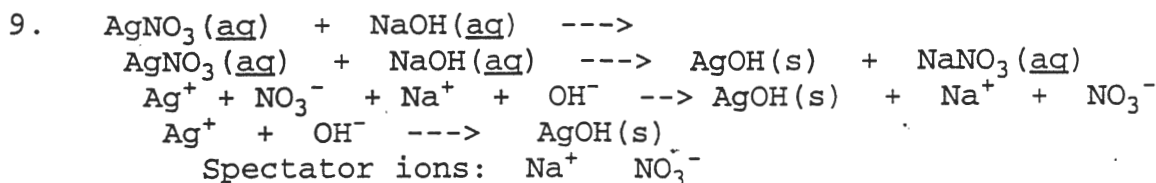
1. Complete and Balance each reaction.
2. Write out the overall IONIC equation.
3. Write out the NET IONIC equation.
4. Identify the SPECTATOR IONS.

Single Replacement Reactions:

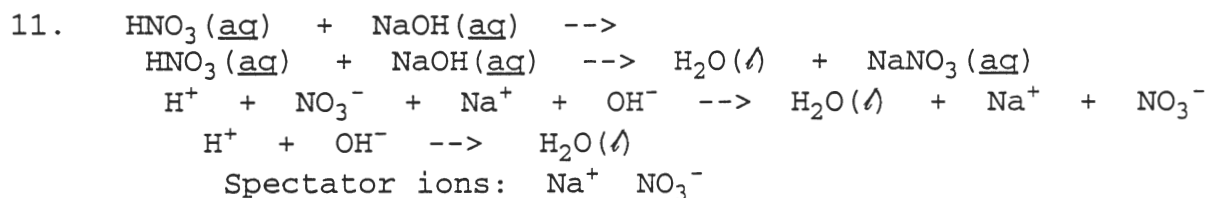
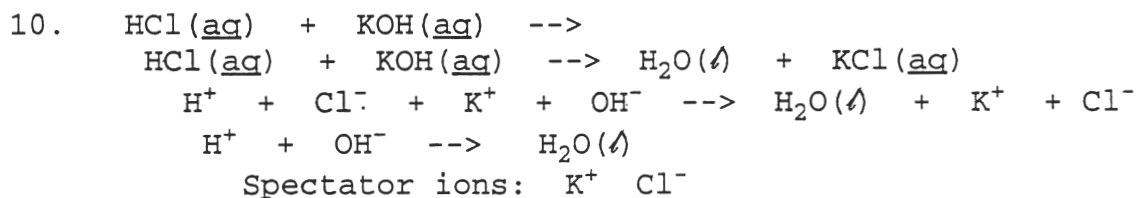


Metathesis Reactions (Double Replacement): *solubility rules are used.*

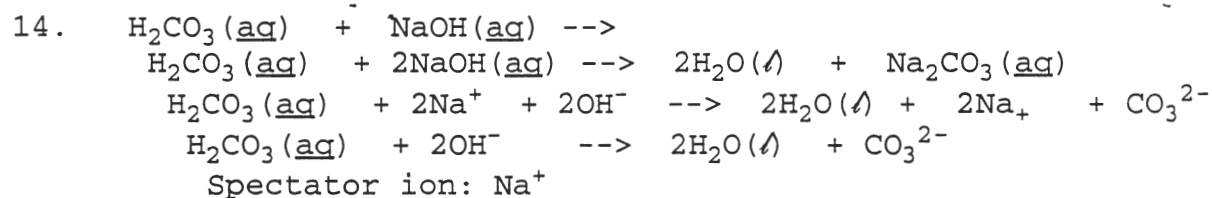
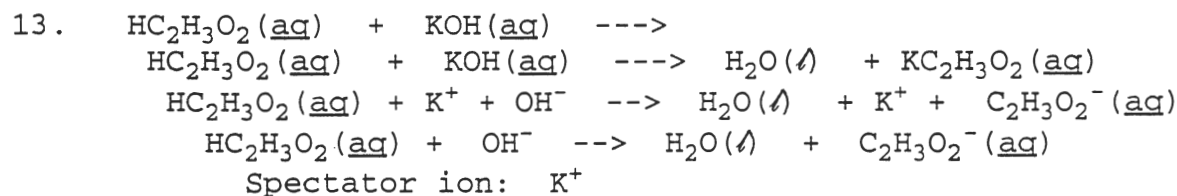
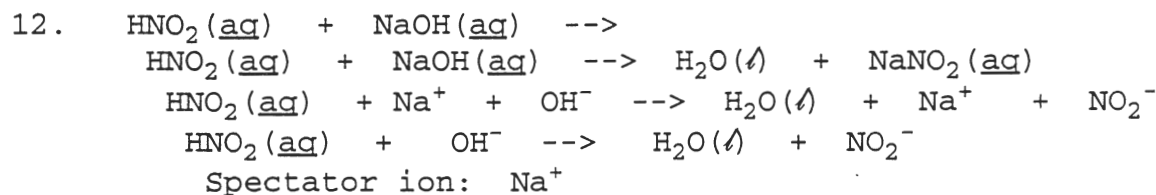




Acid - Base Reactions:



Weak Acid - Strong Base:



All Types:

