

Name: \_\_\_\_\_ Date: \_\_\_\_\_

1. The systematic name for  $\text{BaH}_2$  is
  - A) barium dihydrogen.
  - B) barium dihydride.
  - C) barium(II) hydrate.
  - D) barium dihydrate.
  - E) barium hydride.
  
2. What is the name of the compound whose formula is  $\text{Al}_2(\text{SO}_4)_3$ ?
  - A) aluminum sulfide
  - B) aluminum sulfite
  - C) aluminum sulfate
  - D) aluminum persulfate
  - E) dialuminum tri(sulfur tetraoxygen)
  
3. The correct name for  $\text{FeO}$  is
  - A) iron oxide.
  - B) iron(II) oxide.
  - C) iron(III) oxide.
  - D) iron monoxide.
  - E) iron(I) oxide.
  
4. What is the formula for calcium nitride?
  - A)  $\text{Ca}_3\text{N}_2$
  - B)  $\text{Ca}_2\text{N}_3$
  - C)  $\text{CaNO}_2$
  - D)  $\text{Ca}(\text{NO}_2)_2$
  - E)  $\text{Ca}(\text{NO}_3)_2$
  
5. The chemical formula for iron(III) sulfide is
  - A)  $\text{Fe}_3(\text{SO}_3)_2$ .
  - B)  $\text{Fe}_3(\text{SO}_4)_2$ .
  - C)  $\text{Fe}_2\text{S}_3$ .
  - D)  $\text{Fe}_2(\text{SO}_3)_3$ .
  - E)  $\text{Fe}_2(\text{SO}_4)_3$ .

6. The formula for copper(II) phosphate is
- A) CuPO<sub>4</sub>.
  - B) Cu<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>.
  - C) Co<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub>.
  - D) Cu<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub>.
  - E) Co<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>.
7. Choose the name–formula pair that does not match.
- A) sodium sulfite, Na<sub>2</sub>SO<sub>3</sub>
  - B) calcium fluoride, CaF<sub>2</sub>
  - C) potassium permanganate, K<sub>2</sub>MnO<sub>4</sub>
  - D) aluminum oxide, Al<sub>2</sub>O<sub>3</sub>
  - E) iron(III) oxide, Fe<sub>2</sub>O<sub>3</sub>
8. Choose the name–formula pair that does not match.
- A) ammonium hydrogen carbonate, NH<sub>4</sub>CO<sub>3</sub>
  - B) sodium chlorite, NaClO<sub>2</sub>
  - C) calcium hydride, CaH<sub>2</sub>
  - D) nitric acid, HNO<sub>3</sub>
  - E) calcium hydroxide, Ca(OH)<sub>2</sub>
9. The formula for potassium carbonate is
- A) P<sub>2</sub>CO<sub>3</sub>.
  - B) K<sub>2</sub>CO<sub>3</sub>.
  - C) Po<sub>2</sub>CO<sub>3</sub>.
  - D) P<sub>2</sub>C.
  - E) K<sub>2</sub>C.
10. The formula for magnesium nitride is
- A) MgN.
  - B) MgNO<sub>2</sub>.
  - C) Mg<sub>2</sub>N<sub>3</sub>.
  - D) Mg(NO<sub>2</sub>)<sub>2</sub>.
  - E) Mg<sub>3</sub>N<sub>2</sub>.
11. The correct name for LiCl is
- A) lithium monochloride.
  - B) lithium(I) chloride.
  - C) monolithium chloride.
  - D) lithium chloride.
  - E) monolithium monochloride.

12. The formula for chloric acid is
- A)  $\text{HClO}_4$ .
  - B)  $\text{HClO}_3$ .
  - C)  $\text{HClO}$ .
  - D)  $\text{HClO}_2$ .
  - E)  $\text{HCl}$ .
13. Which name–formula pair is incorrect?
- A)  $\text{H}_2\text{SO}_3$ , sulfurous acid
  - B)  $\text{HClO}_2$ , chlorous acid
  - C)  $\text{H}_2\text{CO}_3$ , carbonic acid
  - D)  $\text{HBr}$ , hydrobromic acid
  - E)  $\text{H}_3\text{P}$ , phosphoric acid
14. Which name–formula pair is incorrect?
- A) magnesium sulfate heptahydrate,  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$
  - B) strontium nitride,  $\text{Sr}_3\text{N}_2$
  - C) titanium(IV) carbide,  $\text{TiC}$
  - D) dinitrogen tetroxide,  $\text{N}_2\text{O}_4$
  - E) hypochlorous acid,  $\text{HClO}_2$
15. Which of the following equations is not balanced?
- A)  $2\text{NaCl} + \text{Pb}(\text{NO}_3)_2 \rightarrow \text{PbCl}_2 + 2\text{NaNO}_3$
  - B)  $\text{Fe}_3\text{O}_4 + 3\text{CO} \rightarrow 3\text{Fe} + 3\text{CO}_2$
  - C)  $(\text{NH}_4)_2\text{Cr}_2\text{O}_7 \rightarrow \text{N}_2 + 4\text{H}_2\text{O} + \text{Cr}_2\text{O}_3$
  - D)  $2\text{Sb}_2\text{OS}_2 + 10\text{O}_2 \rightarrow 2\text{Sb}_2\text{O}_5 + 4\text{SO}_3$
  - E)  $\text{C}_{12}\text{H}_{22}\text{O}_{11} + 12\text{O}_2 \rightarrow 12\text{CO}_2 + 11\text{H}_2\text{O}$

## **Answer Key**

1. E
2. C
3. B
4. A
5. C
6. B
7. C
8. A
9. B
10. E
11. D
12. B
13. E
14. E
15. B